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nebulae luminosity functions (figures 18, 26 in ref. 26), if the population is typically older than 1 Gyr, then $t_{300} \le 2$ Gyr. We therefore adopt $t \le 3$ Gyr as a limit for most of the observed planetary nebulae in the four galaxies. This indicates an association with the "young" simulated stars, and that the mergers of gaseous disks are relevant to those ellipticals showing planetary nebulae. A caveat is the apparent relative invariance of the planetary-nebula luminosity function between galaxies, seemingly independent of signs for a recent major merger. When there are no such signs, the observed planetary nebulae may be the signature of recent minor mergers, which are expected to produce similar effects.

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Isotope-induced partial localization of core electrons in the homonuclear molecule N2

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Because of inversion symmetry and particle exchange, all constituents of homonuclear diatomic molecules are in a quantum mechanically non-local coherent state; this includes the nuclei and deep-lying core electrons. Hence, the molecular photoemission can be regarded as a natural double-slit experiment': coherent electron emission originates from two identical sites, and should give rise to characteristic interference patterns2. However, the quantum coherence is obscured if the two possible symmetry states of the electronic wavefunction ('gerade' and 'ungerade') are degenerate; the sum of the two exactly resembles the distinguishable, incoherent emission from two localized core sites. Here we observe the coherence of core electrons in N₂ through a direct measurement of the interference exhibited in their emission. We also explore the gradual transition to a symmetry-broken system of localized electrons by comparing different isotopesubstituted species—a phenomenon analogous to the acquisition of partial 'which-way' information in macroscopic double-slit experiments3.

With respect to molecular inversion symmetry, the electronic wavefunctions of homonuclear diatomic molecules can be described as symmetry-adapted linear combinations of the corresponding atomic wavefunctions a and b, a situation actually realized by imposition of a fixed phase between them. For the core electrons these symmetry-adapted wavefunctions \(\Psi \), both gerade (g) and ungerade (u), can be written as:

$$\Psi_g = 1/\sqrt{[2(1+S)]} \times [\Psi_a(r) + \Psi_h(r)]$$
 and

$$\Psi_{s} = 1/\sqrt{[2(1-S)]} \times [\Psi_{t}(r) - \Psi_{b}(r)]$$

respectively, where the phases of the two orbitals in Ψ_n differ by π and the overlap-integral is given by $S = \int \Psi_t(r)\Psi_b(r)dr$. The corresponding molecular orbitals for K-shell electrons are designated as la, and la,

The experimental fingerprint of the coherence of photoelectron emission from the log and log states in molecular nitrogen is the angular distributions in the molecular frame. These are predicted to exhibit characteristic differences between the two symmetry states, particularly in their nodal structure, which reflect the angular-momentum-dependent partial wave composition of the photoelectron wavefunction. In the non-local coherent case, this composition should be strictly governed by parity selection rules for the gerade and ungerade final core hole states, giving rise to purely odd and even angular momenta in the corresponding partial waves of

the Io, and Io, photolines. The well-known alternating intensities in rotational spectra resulting from the symmetry that must be imposed on the nuclear spin function to make the complete eigenfunction of the molecule either symmetric or antisymmetric could be viewed as the nuclear analogue of this electronic selectivity behaviour. Replacing one particle in such a system by a different one leads to a complete breakdown of the symmetry properties of the

The showcase example for such complete change in behaviour is the rotational structure in homonuclear diatomic molecules under isotope substitution mentioned above. Here the symmetry selection rules totally collapse and all forbidden or suppressed rotational transitions become equally allowed4. On the other hand, the electronic charge distribution in such a molecule is virtually unchanged by isotope substitution. Indeed, according to the Born-Oppenheimer approximation with its complete decoupling of nuclear and electronic motion, no change should occur in the electronic wavefunction of a hetero-isotopic homonuclear molecule. Known violations of the symmetry rules for the ground vibrational and electronic state of homonuclear diatomic molecules are miniscule', and a symmetry breakdown has been observed for highly excited states only 60. Hence, any observable isotope effects on the electronic wavefunction for core electrons, the key element for chemical and structural analysis of matter10, might seem quite unexpected.

Here we show that inversion symmetry indeed causes non-local, coherent behaviour of the core electron photoemission from homonuclear diatomic molecules such as N2 (ref. 11 and references therein). Our results show that this non-locality changes in a continuous way into partially localized behaviour, if inversion symmetry violations such as isotope substitution are induced.

The experiments were performed with vacuum ultraviolet synchrotron radiation from beamline BW3 of HASYLAB at DESY and beamline UE56/2-PGM1 and UE56/1-PGM at BESSY using a set of electron time-of-flight spectrometers in combination with an ion time-of-flight spectrometer with a position-sensitive anode (Fig. 1). This set-up makes it possible to determine all photoelectron and fragment ion momenta in coincidence, yielding, in the axial recoil approximation, the photoelectron angular distribution of fixed-inspace molecules 12,13. Because the N1:N(1s)-doublet splitting of less than 100 meV (ref. 14) had to be resolved while data were acquired over several days, these measurements required extremely high-energy resolution of both the beamline (40 meV) and our set-up (60 meV) as well as a very high photon beam stability, particularly regarding the photon energy (10⁻⁵). Unresolved spectra would provide the sum of

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the gerade and ungerade photoemission channels15-19, which displays no effect of nonlocality of the core electrons.

Figure 2 shows the resolved molecule-frame angular distributions of the 10, and 10, photoelectrons, together with the angular difference arising from the interfering molecular charge distributions. Less characteristic, but still symmetry-specific behaviour occurs also in the non-coincident laboratory-frame angular distribution characterized by the photoelectron angular distribution parameter β (ref. 20) shown in Fig. 3. The distinction is particularly pronounced in the region of the trapped f-wave resonance at 9 eV above the N₅:N(1s) threshold because of the dominance of this f-partial continuum wave in only one (gerade) of the two photoelectron channels21. Note that the predicted crossover of the two gerade and ungerade angular distribution curves marks the onset of an oscillation driven by the spatial interference of the outgoing

We investigated the transition to the symmetry-broken system by comparing the naturally most abundant (4.14N, nitrogen molecule to two different isotopomers singly substituted ^{16,15}N₂ and doubly substituted ^{15,15}N₂ (both 99% purity). Using the electron spectrometer in a non-coincident mode, that is, without detecting the corresponding fragment ions, we studied the effect of isotope substitution on the photoelectron spectrum. These effects are best illustrated in the ratio of the photoelectron spectra of normal and substituted nitrogen. Figure 4 shows the ratio between the 1s photoelectron spectra of normal $^{14,15}N_2$ and the isotopomer $^{14,15}N_2$ detected at the β -independent 'magic angle' θ_m (54.7° with

respect to the electric vector of the ionizing radiation as shown in Fig. 3), where the measured photoelectron intensity is directly proportional to the partial cross-section30, as well as at 00 where the β -dependence is largest (see Fig. 3). The experimental data (purple circles) are shown together with a model calculation of the vibrational effect due to the mass-dependence of the vibrational

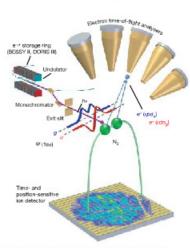


Figure 1 | Experimental set-up for a photoelectron-fragment ion coincidence experiment at a synchrotron radiation source. (For further details, see ref. 13.)

constant (dashed black line in Fig. 4c and d), which causes nuclear-

dynamics-dependent spectral changes. We attribute the variation of the cross-sections and angular distributions beyond this behaviour to the breaking of inversion symmetry in the singly substituted species, which results in a partial localization of the core hole. Whereas the inversion symmetry of N₂ is preserved in the doubly substituted species ^{13,15}N₂, the electronic wavefunction in 14,15 N2 is slightly modified owing to the broken symmetry of the singly substituted molecule, where the centre of symmetry riav of the electric charges midway between the two atoms no longer coincides with the centre of mass $r_{\rm on}$ (ref. 8, 9). The wavefunctions in the molecule with broken inversion symmetry lose their character as parity eigenfunctions and can be described by linear combinations of the original gerade and ungerade wavefunctions. This mixing leads to greater similarity in the cross-sections and angular distributions, resulting in an isotope-induced effect in the range of a few per cent.

Two questions arise in this context: How may we understand the size of the observed effect, and why has it not been seen before in the photoelectron spectra of any homonuclear diatomic molecule? Both questions are closely related and require energy considerations related to the so-called diagonal and nondiagonal asymmetries in a bipolar system. (Here the terms 'diagonal' and 'nondiagonal' refer to contributions to the total hamiltonian of the system, which appear as diagonal and off-diagonal elements, respectively, when the origin of the coordinate system is equidistant from the two nuclei rather than at the centre of mass⁸.) In $^{14,15}N_{2}$, the centre of mass r_{cm} is shifted away from the inversion centre rine by 1.7% of the bond length, resulting in 3.5% asymmetric motion per nucleus due to the

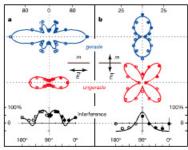


Figure 2 | Symmetry resolved photoelectron distributions in the molecule frame. Molecule frame photoelectron angular distributions (MPADs) for the gerade (upper panel) and augerade (middle panel) N(1s) core photoelectron emission of N_2 at a photon energy of $h\nu = 419$ eV measured in the plane perpendicular to the light propagation direction for molecules oriented parallel (a) and perpendicular (b) to the light polarization vector through selection by an ion momentum resolving imaging detector. The fractional interference angular pattern shown in the lower panel are the differences between the gerade and ungerade MPADs divided by their respective sums (g = u)/(g + u). The open circles are the mirror images of the measured data points (full circles), which are obtained by a least-squares fit of the coincident spectra. The error bars reflect the statistical uncertainty (s.d.) of the fit. The solid lines are predictions for non-local, coherent electron emission calculated in the partially relaxed core Hartree-Fock (RCHF) approximation" shown on a relative scale in arbitrary units marked at the upper margin, but unscaled with respect to each other. Note that the sensitivity of the measurements regarding the difference is considerably reduced at joint nodal points because of the low count rate in both

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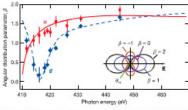


Figure 3 | Photoelectron angular distribution parameter β for N₂:N(1s) electron emission in the photon-energy range 410-450 eV. The symbols represent the experimental data for the gerade (blue diamonds) and inverside (red squares) state. The error bars reflect the statistical and calibrational error. The results are compared to calculations in the partially RCHF approximation". The dashed line shows the results for the perade state, whereas the solid line represents the calculation for the ungerade state Our RCHF curves are generally in good agreement with the data obtained by the Kohn-Sham density functional theory" and the random phase approximation" approach, in particular for the ungerade state (N. A. Cherepkov, personal communication).

relationship $r_{co} = (m_{15} - m_{14})/(m_{15} + m_{14}) \times r_{inc}$, where m_{14} and m_{15} are the masses of ¹⁴N and ¹⁵N, which determines the time during which the electronic wave packet experiences asymmetric motion of the nuclei with respect to the inversion centre. This is similar to the inverse effect of symmetry restoration by detuned excitation in resonant inelastic X-ray scattering²². In dissociating systems, a loss of symmetry has been observed for autoionization²³ and resonant Auger24 lines owing to the Doppler shift during the emission

A system of identical non-overlapping (that is, strongly localized) particles gives rise to completely degenerate gerade and ungerade states. Their underlying symmetry character is inaccessible to experimental exploration by photoelectron spectroscopy because the incoherent sums of gerade and ungerade states and of left- and right-hand states are, by definition, identical. A minimal delocalization resulting in a non-zero overlap between the two core orbitals is required to force a separation between the two symmetry-adapted states of the order of their natural lifetime width. Although this is just the case for N2:N(1s) photoemission, other core level photoelectron spectra of diatomic homonuclear molecules do not fulfil this

In contrast to core levels, the gerade and ungerade splitting for valence levels is very large (in the range of several electron volts) owing to the strong delocalization of most valence electrons. In fact, all ground states are of gerade symmetry—the corresponding ungerade state is unoccupied. One may view the core level splitting in N, as being caused by a core valence-coupling-induced tunnelling rate giving rise to a roughly 15% probability that the electrons from one site will be at the other atomic site25. This tunnelling stabilizes the non-local, coherent character of the electronic state against asymmetric left/right, or, in our terminology, nondiagonal distortions such as a shift of the centre of mass away from the inversion centre, which, for small distortions, may be treated as a perturbation. With regard to localization, it is the ratio of the two associated energies that determines the size of isotope-induced effects in the photoelectron spectra of diatomic homonuclear molecules, analogous to the role of tunnelling versus correlation energy in a superfluid/Mott insulator transition³

Of all the energy effects caused by isotope substitution, we find that only one, the vibrational motion giving rise to an asymmetry energy of the order of several millielectronvolts, is of importance. All other effects, particularly hyperfine perturbations which cause gerade/ungerade symmetry breaking in highly excited states^{6,7} and isotope shifts inducing predissociation in isotopomers⁵³, are in the microelectronvolt range. The observed effect can therefore be viewed as the diatomic analogue of symmetry breaking by vibronic coupling in triatomic molecules27 owing to the asymmetric part of the vibrational motion (in the range of 10 meV). Comparing this 10-meV fraction of the vibrational energy with the gerade/ungerade splitting of 100 meV yields a gerade/ungerade mixing coefficient

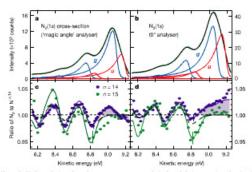
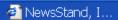


Figure 4 | Isotope effect on N₂. a, b, High-resolution photoelectron spectrum of $^{14,19}N_2$ recorded at a photon energy of $\hbar\nu=419$ eV at the β -independent 'magic angle' θ_{m} (left) and at 0° with respect to the polarization vector of the ionizing radiation (right). The error bars indicate the typical statistical error. The measured spectrum (green line) in the range of the 1s photoline is shown together with the unconvoluted representation of a least-squares fit of its two symmetry-components gerade (blue) and

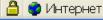
ungerade (red) and their respective vibrational progression up to the third vibrational level. \mathbf{c} , \mathbf{d} , Spectral ratio $^{14,16}\mathrm{N}_3/^{16,15}\mathrm{N}_2$ (purple) compared to $^{14,16}\mathrm{N}_3/^{15,15}\mathrm{N}_2$ (green) for the same angles as above. The solid lines are model calculations which include vibrational effects and, for ^{14,15}N₂, the effect of symmetry breaking on the cross-sections. For ^{14,15}N₂, the dashed line shows a model calculation of the vibrational effect only.









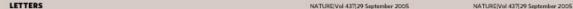






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 $[\Delta m/(m_{14}+m_{15})]/\Delta E_m$ of 10%. To obtain the resulting relative change of the intensities in a first approximation, the square of the mixing coefficient must be multiplied by the normalized intensity difference of the genade and ungenule channels $(I_x - I_n)/(I_x + I_n)$. At the photon energy considered here, this results in an estimated relative change of the cross-section of the order of 1%, which is consistent with the magnitude of the experimentally observed crosssection effect (Fig. 4c). The angle-dependent effect is enlarged owing to the role of the phase shifts between the photoelectron partial waves for all other emission directions besides the magic angle. It is worth mentioning that the size of the observed effect is still too small to be detected unambiguously in coincident measurements as shown in

This analysis also explains why this effect has never been observed in valence photoionization, where the fractional size of the effect is more than two orders of magnitude smaller and where it must be measured on an absolute scale because there are no close-lying gerade and ungerade lines displaying effects in opposite directions. Such small absolute changes of less than 10-4 are inaccessible to photoelectron spectroscopy at present.

In summary, we have shown here that the inversion symmetry of a system indeed causes non-local, coherent behaviour of the otherwise localized core holes in homonuclear diatomic molecules such as No. This non-locality of the electron emission and the remaining core hole is neither conserved nor completely destroyed by a distinct symmetry distortion such as isotope substitution, but instead changes in a continuous way into partially localized behaviour owing to the gradual breakdown of inversion symmetry, as reflected by the loss of interference and parity mixing of the outgoing photoelectron waves. This isotope effect on the electronic structure of a diatomic molecule, probed here by photoelectron spectroscopy. is the first experimentally observed effect of its kind, to our

The continuous nature of this transition, of which we have seen just the onset, makes it possible to control the character of a quantum state from either local or non-local by applying distinct forces that either stabilize or destabilize the non-locality. This knowledge might be useful in other systems such as double quantum dots, which are envisaged as the future building blocks of quantum gates^{26,29}

With the advent of free-electron lasers (FEL)30, which will permit time-resolved pump-probe experiments in the vacuum ultraviolet region, new experiments will become feasible that can probe transitions between complete localization of the electrons on individual atomic sites and complete non-localization over identical sites in analogy to the studies of coherence reported in reference'. To this end, we envision an experiment where a nitrous oxide (N.O) molecule is broken into an O and a N2 fragment by an initial light pulse and the core photoionization of the N₂ fragment is then probed for various time delays as the oxygen moves further and further away. With increasing distance, the emission characteristics should change from the localized, incoherent case of N₂O—where the two nitrogen atoms are distinct because of the chemical shift induced by the oxygen—to the non-local, coherent case of No.

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The model calculations shown in Fig. 4c and d are based on the results of a leastsquares fit of the 14,14N2 spectrum. The change of the vibrational constant is simulated by decreasing the energy spacing between the vibrational components by 5 meV, as predicted by a harmonic oscillator model. The symmetry-induced cross-section effect is modelled by further changing the intensity of the genule and ungerside components. Changes to the Franck-Condon factors caused by the increased reduced mass of the substituted molecule are also included. However, they are found to be relatively small compared to the vibrational and crosssectional effects. The increased mass leads to a decrease of the vibrational energy and therefore to pronounced oscillations in the intensity ratio, which coincide with the position of the vibrational progression. This vibrational effect should be more pronounced for the doubly substituted species 15,15N2 owing to the larger

change of its vibrational energy resulting from its heavier mass. The model calculation reproduces those oscillations very well, but fails to explain the additional 'wiggle' (shaded area) at the high-energy end of the N2/H4,H ratio, which is visible both in the ratio at the magic angle as well as at 0° with respect to the light polarization. This second effect, which does not appear in the N21515N2 ratio (shown in green), can only be explained by a change in both relative intensity as well as angular distribution of the genede and ungerade components in 14.15N₂ compared to the two other isotopomers. The model calculation including these changes (solid lines) reproduces the experimental data even at the high-energy end of the N::N(1s) photoline. It should be noted that the symmetry-induced effect is supposed to appear in all vibrational components; however, for the higher vibrational components below 8.9 eV, the symmetry effect tends to be masked by the vibrational effect. The pure symmetry effect is observed only in the lowest vibrational component above

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